

## MONOTERPENE GLUCOSIDES FROM *BERCHEMIA RACEMOSA*

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**Key Word Index**—*Berchemia racemosa*; Rhamnaceae; monoterpene glucoside; bornanediol glucoside; angelicoidenol glucoside; isoarborinol; 2D-NMR

**Abstract**—Two monoterpene glycosides isolated from the stem of *Berchemia racemosa*, have been characterized as (+)-angelicoidenol-2-*O*- $\beta$ -D-glucopyranoside and (–)-angelicoidenol-2-*O*- $\beta$ -D-glucopyranoside on the basis of chemical and spectral evidence. Isoarborinol was also isolated from the plant.

### INTRODUCTION

From the methanol extract of the stem of *Berchemia racemosa* Sieb. et Zucc., we isolated 2,6-dimethoxybenzoquinone as the physiologically active constituent which inhibits histamine release from rat mast cells induced by compound 48/80 and by Concanavalin A [1]. Recently, we also isolated two new aromatic glycosides, methoxyhydroquinone-4-*O*- $\beta$ -D-glucopyranoside (tachioside) and syringic acid  $\beta$ -D-glucopyranosyl ester, along with three known glycosides, nudiposide, (–)-secoisolaricresinol-9'-*O*- $\beta$ -D-glucopyranoside and methoxyhydroquinone-1-*O*- $\beta$ -D-glucopyranoside (isotachioside) from the butanol-soluble fraction of the methanol extract [2]. We now report on the isolation of the triterpene, isoarborinol (1) from the hexane extract, and two new glucosides (2 and 3) from the butanol-soluble fraction of the methanol extract.

### RESULTS AND DISCUSSION

Compound 1 was identified as isoarborinol by direct comparison of its acetate (1a) with an authentic sample. The physical data of the oxidation product, arborinone (1b) also matched the literature values. The  $^{13}\text{C}$  NMR data of 1a and 1b also supported the structures.

Compound 2,  $\text{C}_{16}\text{H}_{28}\text{O}_7$ ,  $[\text{M} + \text{Na}]^+ m/z$  355.1737 (calc. 355.1733) gave the pentaacetate,  $\text{C}_{26}\text{H}_{38}\text{O}_{12}$ ,  $[\text{M}]^+ m/z$  542. Methanolysis followed by GC showed the presence of glucose (converted to its TMS ether for GC). The  $^{13}\text{C}$  NMR spectrum ( $\text{C}_5\text{D}_5\text{N}$ ) of 2 showed the signals of six  $\beta$ -glucopyranosyl carbons, three methyl groups, two  $\text{CH}_2$  groups, one  $\text{CH}$  group, two quarternary carbons, and two  $\text{CH-OH}$  groups. There were no signals for  $\text{C}=\text{C}$  groups (Table 1).

The three methyl groups gave three singlet peaks in the  $^1\text{H}$  NMR spectrum. These data suggested that the aglycone moiety of 2 ( $\text{C}_{10}\text{H}_{18}\text{O}_2$ ) was a bicyclic monoterpene diol. The chemical shifts of the aglycone moiety were very close to the reported value of angelicoidenol (4)

Table 1.  $^{13}\text{C}$  NMR data and glucosylation shift ( $\Delta\delta$  in parentheses) of compounds 2 and 3 ( $\text{C}_5\text{D}_5\text{N}$ , 67.5 MHz)

C	4*	2 (2-4)	3 (3-4)
1	50.8	50.9 (+0.1)	50.4 (–0.4)
2	75.0	85.2 (+10.2)	82.9 (+7.9)
3	37.1	35.8 (–1.4)	34.2 (–2.9)
4	53.7	53.4 (–0.3)	53.4 (–0.3)
5	75.0	74.8 (–0.2)	74.9 (–0.1)
6	39.6	40.1 (+0.5)	40.1 (+0.5)
7	47.9	47.6 (–0.3)	48.1 (+0.2)
8	21.8	21.3 (–0.5)	21.4 (–0.4)
9	20.2	20.2 (0.0)	20.3 (+0.1)
10	13.5	13.9 (+0.3)	13.6 (+0.1)
G1		106.2 (+7.4)†	103.6 (+4.8)†
G2		75.5	75.3
G3		78.6	78.7
G4		71.7	71.8
G5		78.2	78.3
G6		62.9	62.9

\* Data taken from ref. [3].

† Difference from free  $\beta$ -D-glucose.

[3] except for those of C-2 and C-3. The chemical shift differences ( $\Delta\delta = \delta_{\text{glucoside}} - \delta_{\text{alcohol}}$ ) of these carbons being +10.2 and –1.4, respectively.

In the  $^{13}\text{C}$ - $^1\text{H}$  COSY spectrum, all of the protons were correlated to the appropriate carbon atoms. The  $^1\text{H}$ - $^1\text{H}$  COSY spectrum revealed the vicinal relationship of H-2 $\beta$  ( $\delta$  4.18) and H-3 $\beta$  ( $\delta$  2.38, *ddd*,  $J = 5, 10$  and  $14$  Hz) with H-3 $\alpha$  ( $\delta$  1.48, *dd*,  $J = 3, 14$  Hz); H-3 $\beta$  also correlated to H-4 $\alpha$  ( $\delta$  1.91, *d*,  $J = 5$  Hz). The other correlation was observed for H-5 $\alpha$  ( $\delta$  4.23) and H-6 $\beta$  ( $\delta$  1.74, *br d*,  $J = 13$  Hz) with H-6 $\alpha$  ( $\delta$  2.99, *dd*,  $J = 8$  and  $13$  Hz). Weak but significant coupling was observed between H-6 $\beta$  and H-2 $\beta$ . The fact that H-4 coupled with only one proton can be explained by the  $90^\circ$  dihedral angles between H-4 $\alpha$  and H-3 $\alpha$ , and H-4 $\alpha$  and H-5. All NMR data satisfy the anticipated bornan-2,5-diol skeleton.

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Table 2.  $^1\text{H}$  NMR spectral data of compounds **2–4** ( $\text{C}_5\text{D}_5\text{N-TMS}$ , 270 MHz for **2** and **3**, 90 MHz for **4**\*)

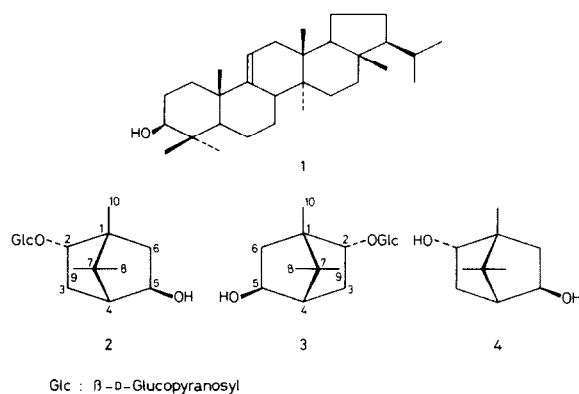
H	<b>4</b> *	<b>2</b>	<b>3</b>
2 $\beta$	4.50 <i>ddd</i>	4.18 <i>br d</i>	4.38 <i>br d</i>
(exo)	(9.5, 3.5, 2)	n.d.†	(9)
3 $\alpha$	1.15 <i>dd</i>	1.48 <i>dd</i>	1.43 <i>dd</i>
(endo)	(13.5, 3.5)	(14, 3)	(13.5, 3.5)
3 $\beta$	2.20–2.50 <i>ddd</i>	2.38 <i>ddd</i>	2.31 <i>ddd</i>
(exo)	(13.5, 9.5, 5)	(14, 10, 5)	(14, 9, 5)
4 $\alpha$	1.95 <i>d</i> (5)	1.91 <i>d</i> (5)	1.96 <i>d</i> (5)
5 $\alpha$	4.19 <i>dd</i>	<i>ca</i> 4.23	4.30 <i>dd</i>
(endo)	(8.5, 3)		(8, 3)
6 $\alpha$	2.8–3.05 <i>dd</i>	2.99 <i>dd</i>	2.95 <i>dd</i>
(endo)	(13, 8.5)	(13, 8)	(13, 8)
6 $\beta$	1.72 <i>dd</i>	1.74 <i>br d</i>	1.72 <i>br d</i>
(exo)	(13, 3)	(13)	(13)
8 (Me)	1.40 <i>s</i> ‡	1.39 <i>s</i>	1.40 <i>s</i>
9 (Me)	0.90 <i>s</i> ‡	0.85 <i>s</i>	0.84 <i>s</i>
10 (Me)	1.05 <i>s</i> ‡	1.20 <i>s</i>	1.12 <i>s</i>
1'	Glucosyl moiety	4.93 <i>d</i> (8)	4.90 <i>d</i> (8)
2'		4.03 <i>dd</i> (10, 8)	4.01 <i>dd</i> (9, 8)
3'		4.24 n.d.†	4.25 n.d.
4'		4.26 n.d.	4.27 n.d.
5'		3.95 <i>ddd</i>	3.95 <i>ddd</i>
		(10, 5, 3)	(9, 5, 3)
6'		4.40 <i>dd</i> (12, 5)	4.40 <i>dd</i> (12, 5)
6''		4.53 <i>dd</i> (12, 3)	4.54 <i>dd</i> (12, 3)

Coupling constants in Hz are in parentheses.

\*Data taken from ref. [3].

†Not determined due to overlapping of the signals.

‡Assignment interchanged by our C–H COSY results.



Compound **3**,  $\text{C}_{16}\text{H}_{28}\text{O}_7$ ,  $[\text{M} + \text{Na}]^+$   $m/z$  355.1740 (calc. 355.1733) gave rise to very similar  $^{13}\text{C}$  and  $^1\text{H}$  NMR spectra to those **2** (Tables 1 and 2). Methanalysis of **3** followed by GC showed the presence of glucose (as its TMS ether). These findings suggested that **3** was a stereoisomer of **2**.

In order to establish the position of the glucosyl residue and the absolute structure, the  $^{13}\text{C}$  NMR glucosylation shift was considered. In the case of  $\beta$ -D-glucopyranosides of secondary alcohols, the substitution induced shift values of carbon signals caused by glucosylation

(glucosylation shift) depend on the absolute configuration of the alcohol. The shift value ( $\Delta\delta = \delta_{\text{glucoside}} - \delta_{\text{alcohol}}$ ) of an  $\alpha$ -carbon is *ca* 6–8 ppm for an achiral alcohol and a chiral (*R*)-alcohol, and *ca* 10 ppm for a chiral (*S*)-alcohol. Also, the two  $\beta$ -carbons of the aglycone and the anomeric carbon reflect the absolute stereochemistry [4, 5].

The glucosylation shifts of both compound indicated that the glucosylated position was restricted to the C-2 hydroxyl group, and that the absolute configuration of this position is *S* for **2** [by reference to the reported value of dammaranediol (**5**)] and *R* for **3** [by analogy with ent-dammaranediol (**6**\*)] (Table 1 and Fig. 1) [4].

Although the aglycones of both compounds were not obtained, they should be a pair of enantiomers. This was confirmed by Klyne's rule of glycosylation [6]. The calculated specific optical rotation of **2** and **3** from the values of **4** and methyl  $\beta$ -D-glucoside were  $-11.0^\circ$  and  $-27.5^\circ$ , respectively, which are in good accordance with the observed values,  $-12.6^\circ$  and  $-26.3^\circ$ , respectively. Thus, the structures of two glucosides, **2** and **3** were elucidated to be 2- $\beta$ -D-glucopyranosyl-(+)-angelicoidenol and 2- $\beta$ -D-glucopyranosyl-(–)-angelicoidenol, respectively. It is to be noted that two glucosides with enantiomeric aglycones exist in the same plant.

## EXPERIMENTAL

Mp: uncorr;  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR: 270 and 67.5 MHz, respectively, except when otherwise stated; MS: 75 eV.

**Plant material.** *Berchemia racemosa* Sieb. et Zucc. was collected in the vicinity of Taishaku-kyo, Hiroshima Prefecture, Japan. A specimen is deposited at the Herbarium of Experimental Station of Medicinal Plants, Hiroshima University School of Medicine.

**Extraction and isolation.** Dried stem of *B. racemosa* (2.0 kg) were crushed and extracted with *n*-hexane followed by MeOH. The MeOH extract was suspended in  $\text{H}_2\text{O}$ , and extracted successively with *n*-hexane,  $\text{Et}_2\text{O}$ ,  $\text{EtOAc}$ , BuOH and  $\text{H}_2\text{O}$ . From a

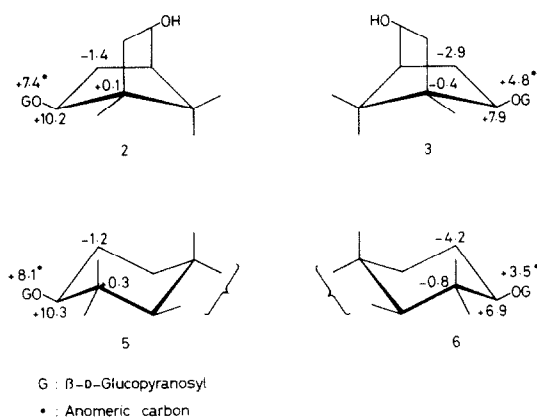


Fig. 1.  $^{13}\text{C}$  NMR glucosylation shifts ( $\Delta\delta$ ) on hydroxyl-bearing carbons, anomeric carbons and carbon atoms adjacent to hydroxyl-bearing carbons for the determination of absolute configuration of aglycones.

\* Actual  $^{13}\text{C}$  NMR data came from  $\beta$ -L-glucoside of **5** [4].

portion (4.1 g) of the combined hexane extract (10.9 g), isoarborinol (**1**) (180 mg) was isolated by silica gel CC ( $C_6H_6$  upto  $C_6H_6$  with 0.5%  $Me_2CO$ ).

The BuOH-soluble fraction (20.2 g) was chromatographed on highly porous polymer, Diaion HP-20, developed with  $H_2O$  containing the 10, 20, 30, . . . 90% MeOH and 100% MeOH). The 50% MeOH eluent was subjected to silica gel CC ( $CHCl_3$ -MeOH- $H_2O$ ) and DCC ( $CHCl_3$ -MeOH- $H_2O$ , 5:6:4) followed by DCC ( $CHCl_3$ -MeOH- $H_2O$ -propan-1-ol, 5:6:4:1) again to give a mixture of **2** and **3**. This was subjected to Sephadex LH-20 CC (MeOH), prep. HPLC on RP-18 (MeOH- $H_2O$ ) and silica gel CC (EtOAc-EtOH- $H_2O$ ) to afford compounds **2** (54 mg), and **3** (27 mg).

**Isoarborinol (1)**. Colourless needles from  $C_6H_6$  mp 308–310°, lit. [7] mp 295–296°. MS  $m/z$  (rel. int.): 426  $[M]^+$  (20), 411 (30), 393 (6), 259 (20), 43 (100); IR  $\nu_{max}^{KBr}$   $cm^{-1}$ : 3230, 2930, 2860, 1470, 1385, 1375, 1031.

**Isoarborinol acetate (1a)**. Acetylation of **1** (40 mg) in  $C_5H_5N$ - $Ac_2O$  (5 ml each) at 55° for 5 hr afforded **1a** (32 mg), mp 290–292°, lit [7] 287–288°; MS  $m/z$ : 468, 443, 393, 301, 255, 241, 229;  $^1H$  NMR ( $CDCl_3$ ):  $\delta$  0.75 (3H, s), 0.76 (3H, s), 0.80 (3H, s), 0.83 (3H, d,  $J = 6.6$  Hz), 0.86 (3H, s), 0.89 (3H, s), 0.90 (3H, d,  $J = 6.6$  Hz), 1.05 (3H, s), 4.50 (1H, m), 5.23 (1H, d,  $J = 6.1$  Hz); identical with authentic spectrum.  $^{13}C$  NMR (in  $CDCl_3$ , 25 MHz):  $\delta$  14.0, 15.3, 16.8, 17.0, 21.4, 22.1, 23.0, 28.2  $\times 2$  ( $9 \times CH_3$ ), 20.2, 21.4, 24.2, 26.6, 28.2, 29.7, 35.7, 36.0  $\times 2$  ( $9 \times CH_2$ ), 30.8, 41.0, 52.1, 52.5, 59.7 ( $5 \times CH$ ), 36.8, 38.1, 38.2, 39.5, 42.9 ( $5 \times qC$ ), 81.0 (CH-O), 114.6 ( $\underline{CH} = C$ ), 148.5 ( $CH = \underline{C} <$ ), 170.8 ( $-\underline{COOMe}$ ).

**Arborinone (1b)**. Oxidation of **1** (80 mg) with 102 mg  $CrO_3$  in  $H_2O$  (10 ml), AcOH (2 ml) and  $C_6H_6$  (40 ml) for 3 hr afforded arborinone (**1b**) (60 mg), mp 219–221°, lit [7], 214–214.5°; IR  $\nu_{max}^{KBr}$   $cm^{-1}$ : 3400, 2930, 2860, 1700, 1655, 1270, 1000;  $^{13}C$  NMR (in  $CDCl_3$ , 25 MHz):  $\delta$  14.0, 15.2, 17.0, 21.6, 22.0  $\times 3$ , 25.5 ( $8 \times Me$ ), 20.1, 22.5, 23.0, 26.2, 28.2, 29.6, 34.8, 35.9, 36.6 ( $9 \times CH_2$ ), 30.8, 41.1, 51.9, 53.2, 59.6 ( $5 \times CH$ ), 35.9, 38.1, 39.3, 42.8, 47.6 ( $5 \times qC$ ), 115.6 ( $\underline{CH} = C$ ), 147.4 ( $C = \underline{C} <$ ), 214.9 ( $C = O$ ).

(+)-**Angelicoidenol-2-O- $\beta$ -D-glucopyranoside (2)**. Amorphous powder,  $[\alpha]_D^{20} = -12.6^\circ$  (MeOH;  $c$  0.65). FABMS (DMSO, glycerol + NaI)  $m/z$  355.1737  $[M + Na]^+$  (calc. 355.1733,  $C_{16}H_{28}O_7$  Na); IR  $\nu_{max}^{KBr}$   $cm^{-1}$ : 3600–3100 (OH), 2900, 1450, 1385, 1360, 1285, 1230, 1160, 1075, 1025;  $^1H$  NMR and  $^{13}C$  NMR; see Tables 1 and 2.

(-)-**Angelicoidenol-2-O- $\beta$ -D-glucopyranoside (3)**. Amorphous powder,  $[\alpha]_D^{20} = -26.3^\circ$  (MeOH;  $c$  1.04). FABMS (DMSO, glycerol + NaI)  $m/z$  355.1740  $[M + Na]^+$  (calc. 355.1733,  $C_{16}H_{28}O_7$  Na);  $^1H$  NMR and  $^{13}C$  NMR; see Tables 1 and 2.

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